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Phosphorus, Sulfur, and Silicon and the Related Elements

Publication details, including instructions for authors and subscription information: http://www.informaworld.com/smpp/title~content=t713618290

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Online publication date: 27 October 2010

To cite this Article Weber, Lothar , Meyer, Marco and Quasdorff, Beate(2002) 'Reactivity of Phosphaalkenes toward Carbene Complexes', Phosphorus, Sulfur, and Silicon and the Related Elements, 177: 6, 1571 - 1574

To link to this Article: DOI: 10.1080/10426500212215 URL: http://dx.doi.org/10.1080/10426500212215

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Phosphorus, Sulfur and Silicon, 2002, Vol. 177:1571–1574 Copyright © 2002 Taylor & Francis 1042-6507/02 \$12.00 + .00

DOI: 10.1080/10426500290092794



REACTIVITY OF PHOSPHAALKENES TOWARD CARBENE COMPLEXES

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(Received July 29, 2001; accepted December 25, 2001)

Reaction of phosphaalkenes $RP = C(NMe_2)_2$ (R = t-Bu, Me_3Si), featuring an inverse distribution of electron density about the P-C double bond, with Fischer carbene complexes $[(CO)_5M = C(OEt)Ar]$ (Ar = Ph, $2\text{-}MeC_6H_4$, $2\text{-}MeOC_6H_4$, M = Cr, W) afforded a mixture of complexes $[(CO)_5M[P(R) = C(NMe_2)_2]]$ and $[(CO)_5M[P(R) = C(OEt)Ar]]$. The treatment of phosphaalkene $HP = C(NMe_2)_2$ with compound $[(CO)_5W = C(OEt)(2\text{-}MeOC_6H_4)]$ gives rise to the formation of an (E/Z)-mixture of $[(CO)_5W[P(CH(NMe_2)_2) = C(OEt)(2\text{-}MeOC_6H_4)]]$.

Keywords: Carbene ligand; chromium; insertion; phosphaalkenes; tungsten

The vast majority of phosphaalkenes feature an electron distribution $P^{\delta+}C^{\delta-}$ at the P–C double bond, as would be anticipated from the different electronegativities of carbon (2.5) and phosphorus (2.1). Recently we have become interested in phosphaalkenes with an inverse polarity of the P–C double bond, which means that negative charge is accumulated at the phosphorus atom and a deficiency of charge resides on the carbon atom. This situation is given when one or two amino substituents are linked to the carbon atom of the P=C unit.

Formula **B** (Figure 1) underlines the nucleophilic character of the phosphorus atom in such species. Here we give an account on the reactivity of phosphaalkenes of the type $RP=C(NMe_2)_2$ ($R=Cp^*(CO)_2Fe$, H, tBu, Me_3Si) toward Fischer carbene complexes.

When ferriophosphaalkene **1** was exposed to 2 molar equivalents of the Cr- and W-methyl carbene complexes, the formation of novel ferriophosphaalkene-pentacarbonyl metal complexes **2** and β -amino-alkenyl carbene complexes **3** was observed (Figure 2). The combination of the corresponding phenyl carbene complexes

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FIGURE 1

$$[Fe] - P = C(NMe_2)_2$$

$$1$$

$$[Fe] - P = C(NMe_2)_2$$

$$1$$

$$[Fe] - [Cp*(CO)_2Fe]$$

$$(CO)_5M = C$$

FIGURE 2

$$[W] = C \xrightarrow{\text{OEt}} \frac{\text{RP} = C(\text{NMe}_2)_2}{\text{Ph}} \xrightarrow{\text{RP} = C(\text{NMe}_2)_2} \xrightarrow{\text{RP} = C(\text{NMe}_2)_2} + \underbrace{\begin{array}{c} \text{RP} = C(\text{NMe}_2)_2 \\ \text{Ph} \end{array}}_{\text{Ph}} \xrightarrow{\text{P} = C(\text{NMe}_2)_2} + \underbrace{\begin{array}{c} \text{RP} = C(\text{NMe}_2)_2 \\ \text{Ph} \end{array}}_{\text{Ph}} \xrightarrow{\text{P} = C(\text{NMe}_2)_2} + \underbrace{\begin{array}{c} \text{RP} = C(\text{NMe}_2)_2 \\ \text{Ph} \end{array}}_{\text{Ph}} \xrightarrow{\text{P} = C(\text{NMe}_2)_2} + \underbrace{\begin{array}{c} \text{RP} = C(\text{NMe}_2)_2 \\ \text{Ph} \end{array}}_{\text{Ph}} \xrightarrow{\text{P} = C(\text{NMe}_2)_2} + \underbrace{\begin{array}{c} \text{RP} = C(\text{NMe}_2)_2 \\ \text{Ph} \end{array}}_{\text{P} = C(\text{NMe}_2)_2} + \underbrace{\begin{array}{c} \text{RP} = C(\text{NMe}_2)_2 \\ \text{Ph} \end{array}}_{\text{P} = C(\text{NMe}_2)_2} + \underbrace{\begin{array}{c} \text{RP} = C(\text{NMe}_2)_2 \\ \text{Ph} \end{array}}_{\text{P} = C(\text{NMe}_2)_2} + \underbrace{\begin{array}{c} \text{RP} = C(\text{NMe}_2)_2 \\ \text{Ph} \end{array}}_{\text{P} = C(\text{NMe}_2)_2} + \underbrace{\begin{array}{c} \text{RP} = C(\text{NMe}_2)_2 \\ \text{Ph} \end{array}}_{\text{P} = C(\text{NMe}_2)_2} + \underbrace{\begin{array}{c} \text{RP} = C(\text{NMe}_2)_2 \\ \text{Ph} \end{array}}_{\text{P} = C(\text{NMe}_2)_2} + \underbrace{\begin{array}{c} \text{RP} = C(\text{NMe}_2)_2 \\ \text{Ph} \end{array}}_{\text{P} = C(\text{NMe}_2)_2} + \underbrace{\begin{array}{c} \text{RP} = C(\text{NMe}_2)_2 \\ \text{Ph} \end{array}}_{\text{P} = C(\text{NMe}_2)_2} + \underbrace{\begin{array}{c} \text{RP} = C(\text{NMe}_2)_2 \\ \text{Ph} \end{array}}_{\text{P} = C(\text{NMe}_2)_2} + \underbrace{\begin{array}{c} \text{RP} = C(\text{NMe}_2)_2 \\ \text{Ph} \end{array}}_{\text{P} = C(\text{NMe}_2)_2} + \underbrace{\begin{array}{c} \text{RP} = C(\text{NMe}_2)_2 \\ \text{Ph} \end{array}}_{\text{P} = C(\text{NMe}_2)_2} + \underbrace{\begin{array}{c} \text{RP} = C(\text{NMe}_2)_2 \\ \text{Ph} \end{array}}_{\text{P} = C(\text{NMe}_2)_2} + \underbrace{\begin{array}{c} \text{RP} = C(\text{NMe}_2)_2 \\ \text{Ph} \end{array}}_{\text{P} = C(\text{NMe}_2)_2} + \underbrace{\begin{array}{c} \text{RP} = C(\text{NMe}_2)_2 \\ \text{Ph} \end{array}}_{\text{P} = C(\text{NMe}_2)_2} + \underbrace{\begin{array}{c} \text{RP} = C(\text{NMe}_2)_2 \\ \text{Ph} \end{array}}_{\text{P} = C(\text{NMe}_2)_2} + \underbrace{\begin{array}{c} \text{RP} = C(\text{NMe}_2)_2 \\ \text{Ph} \end{array}}_{\text{P} = C(\text{NMe}_2)_2} + \underbrace{\begin{array}{c} \text{RP} = C(\text{NMe}_2)_2 \\ \text{Ph} \end{array}}_{\text{P} = C(\text{NMe}_2)_2} + \underbrace{\begin{array}{c} \text{RP} = C(\text{NMe}_2)_2 \\ \text{Ph} \end{array}}_{\text{P} = C(\text{NMe}_2)_2} + \underbrace{\begin{array}{c} \text{RP} = C(\text{NMe}_2)_2 \\ \text{Ph} \end{array}}_{\text{P} = C(\text{NMe}_2)_2} + \underbrace{\begin{array}{c} \text{RP} = C(\text{NMe}_2)_2 \\ \text{Ph} \end{array}}_{\text{P} = C(\text{NMe}_2)_2} + \underbrace{\begin{array}{c} \text{RP} = C(\text{NMe}_2)_2 \\ \text{Ph} \end{array}}_{\text{P} = C(\text{NMe}_2)_2} + \underbrace{\begin{array}{c} \text{RP} = C(\text{NMe}_2)_2 \\ \text{Ph} \end{array}}_{\text{P} = C(\text{NMe}_2)_2} + \underbrace{\begin{array}{c} \text{RP} = C(\text{NMe}_2)_2 \\ \text{Ph} \end{array}}_{\text{P} = C(\text{NMe}_2)_2} + \underbrace{\begin{array}{c} \text{RP} = C(\text{NMe}_2)_2 \\ \text{Ph} \end{array}}_{\text{P} = C(\text{NMe}_2)_2} + \underbrace{\begin{array}{c} \text{RP} = C(\text{NMe}_2)_2 \\ \text{Ph} \end{array}}_{\text{P} = C(\text{NMe}_2)_2} + \underbrace{\begin{array}{c} \text{RP} = C(\text{NM$$

FIGURE 3

$$[W] = C \xrightarrow{\text{Me}_3 \text{SiP} = C(\text{NMe}_2)_2} \xrightarrow{\text{Me}_3 \text{Si}} P = C \xrightarrow{\text{NMe}_2} \xrightarrow{\text{NMe}_2} \xrightarrow{\text{NMe}_2} \xrightarrow{\text{NMe}_2} \xrightarrow{\text{NMe}_3 \text{Si}} P = C \xrightarrow{\text{NMe}_3 \text{Si}} P = C \xrightarrow{\text{NMe}_2} \xrightarrow{\text{NMe}_3 \text{Si}} P = C \xrightarrow{$$

FIGURE 4

$$[W] = C \xrightarrow{\text{RP} = C(\text{NMe}_2)_2} \xrightarrow{\text{RP} = C$$

FIGURE 5

with 1 in n-pentane did not give tractable metathesis products. Instead, the dimeric complex $[Cp^*(CO)_2Fe]_2$ (70%) and the adduct $(CO)_5Cr[P\{[Fe]=C(NMe_2)_2\}]$ (10%) were isolated. Obviously, this reaction was dominated by the cleavage of the Fe–P bond of 1.²

Treatment of the tungsten phenyl carbene complex with the metal-free phosphaalkenes $RP=C(NMe_2)_2$ (R=t-Bu, Me_3Si) gave a mixture of the metathesis products **5** and **7** with the complexes **4** and **6** and olefin **8** (Figure 3).³ Unfortunately, **5** and **7** could not be isolated without decomposition.

An *o*-tolyl carbene tungsten complex did not react with *t*-BuP=C(NMe₂)₂; whereas with Me₃SiP=C(NMe₂)₂ complexes **6** and **9** could be isolated as yellow crystals (Figure 4).

The related o-anisyl carbene tungsten complex and the phosphaalkenes RP=C(NMe₂)₂ (R=tBu, Me₃Si) underwent reaction to yield mixtures of the adducts **4** and **6** in addition to the metathesis products **10** and **11** and alkene **12** (Figure 5).

Treatment of HP= $C(NMe_2)_2$ with $(CO)_5W$ =C(OEt)Ar (Ar = Ph, 2-MeC₆H₄) has led to decomposition. With the *o*-anisyl carbene complex, however, the formation of the novel phosphalkene complex **13** as a (E/Z) mixture (1:2) (80% yield) was observed (Figure 6).

[W]=COEt

HP=C(NMe₂)₂

[W] - PCH(NMe₂)₂

$$OCH_3$$

[W] = [W(CO)₅]

FIGURE 6

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